

Preparation and *In Vitro* Evaluation of Self-Nanoemulsifying Drug Delivery System (SNEDDS) Containing Clopidogrel

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ABSTRACT

The oral delivery of lipophilic drugs presents a major challenge due to low aqueous solubility of such compounds. Clopidogrel is a BCS class II prodrug specifically and irreversibly inhibits the platelets aggregation by blocking activation of the glycoprotein IIb /Illa pathway. The chief intention of this work is to develop an orally stable self Nano-emulsifying drug delivery system by evaluating its *in vitro* potential. Components of SNEDDS were assessed by solubility studies on various oils, surfactant, co-surfactants and co solvents. Ternary phase diagrams were constructed to identify area of nanoemulsification for the selected systems. Characterization of SNEDDS was done by Physical method, Droplet size, Zeta potential determination, drug loading capacity, Transmission test, Cloud point measurement and *in vitro* release study. The optimal Formulation consisted of mixture of Drug (13.05%), Acrysol K150 and PEG 400 (1:1) and Capmul MCM NF (17.39%). Droplet size of optimal batch was 22.91 nm with Pdl 0.173.Drug loading capacity was 2 times the Actual dose of CLP (75 mg). Transmission values were above 99% in pH 1.2, Ph 6.8 and distilled water. Cloud point of formulations was above 65°C. *In vitro* release inspection of optimal formulation illustrated a complete release of Clopidogrel from SNEDDS within 15 min. Our study concludes that the SNEDDS shows potential approach for the poorly water soluble drugs including Clopidogrel.

Keywords: SNEDDS, Clopidogrel, Drug loading capacity.

INTRODUCTION

ral route prescribed as the ideal route for chronic drug therapy; yet 40% of new chemical entities exhibit low oral bioavailability and high intra- and inter subject variability due to their poor aqueous solubility.¹ Lipid-based formulations are of good interest in recent years. Lipid carriers such as oils, surfactant, self-emulsifying formulations. emulsions. selfnanoemulsifying systems, self-microemulsifying systems² are having potential to engulf lipophilic drug and most importantly remain inside until reaches to blood circulation. Self nanoemulsifying systems (SNS) are mixtures oil, surfactants, cosurfactants that form fine o/w nanoemulsion when introduced into aqueous phases under gentle agitation.³ various formulation strategies to improve the dissolution and bioavailability have been subjected for BCS Class II drug but amongst them selfnanoemulsifying drug delivery systems (SNEDDS) have established particular interest as a means of enhancing oral bioavailability of poorly absorbed drugs.⁴ SNEDDS spread into fine emulsion droplets inside the gut lumen where drug remains in solution state, evading the dissolution step that intermittently limits the rate of absorption of hydrophobic drugs from the crystalline state.

Clopidogrel is an oral, thienopyridine class antiplatelet agent used to inhibit blood clots in coronary artery disease, peripheral vascular disease, and cerebrovascular disease. Clopidogrel is a prodrug, which action may be related to an ADP receptor on platelet cell membranes. The drug specifically and irreversibly inhibits the P2Y12 subtype of ADP receptor, which is important in activation of platelets and eventual cross-linking by the protein fibrin.⁵ The blockade of this receptor inhibits platelet aggregation by blocking activation of the glycoprotein IIb /IIIa pathway. The IIb/IIIa complex functions as a receptor mainly for fibrinogen. Activation of this receptor complex is the "final common pathway" for platelet aggregation and is important in the cross-linking of platelets by fibrin.

Present study shows preparation of various kind of SNEDDS formulation by performing trial batches of various oils (sunflower oil, soyabean oil, olive oil), surfactants (Acrysol derivatives and other Gattefosse surfactants, Tween 80, Tween 20), cosurfactants (captex 200, captex 300, labrafil M 2125) and cosolvents (PEG 400, Propylene Glycol, ethanol) with drug to assess the ability of drug in solubilized form and to form a transparent solution for better emulsification ratio of Surfactant:Cosurfactant:Cosolvents remains same. Nonionic surfactants with high HLB (HLB = 10) and subsequent hydrophilicity is necessary for the instant creation of oil in water droplets and/or rapid spreading of the formulation in the aqueous environment providing a good dispersing/self-emulsifying performance. The surfactants are amphiphilic in nature have ability to dissolve and solubilize to some extent high quantities of the hydrophobic drug. In SNEDDS generally surfactant of HLB value 8-16 is used.

The prime objective of the investigation is to formulate, optimize and stabilize SNEDDS containing Clopidogrel with suitable surfactants and co-surfactants. To achieve High drug loading capacity with ability to retain drug in solubilized form. Solubility of drug plays a very important



role in dissolution and hence absorption of drug which hypothetically affects its bioavailability.

MATERIALS AND METHODS

Clopidogrel was received as a gift sample from APOTEX Ltd.(Bangalore, Research Pvt. India) Gelucire 44/14,Labrasol, Labrafil M 2125, Capryol 90 were gift samples from Gattefosse (Mumbai, India), Capmul MCM NF, Acconon MC8-2, Captex 200 were kindly provided by Abitech Corporation (Columbus, OH, USA), Acrysol Derivatives (K150, EL135) were generous gift from Coral Pharma (Ahmedabad, India), Tween 80, Tween 20 were kindly gifted by Indoco Pvt. Ltd (Aurangabad, India), PEG 400, Propylene Glycol, Methanol were purchased from S.D. fine chemicals (Mumbai, India) Double distilled water was prepared freshly whenever required. Various oils are purchased as received. All other chemicals were of analytical grade.

Solubility studies

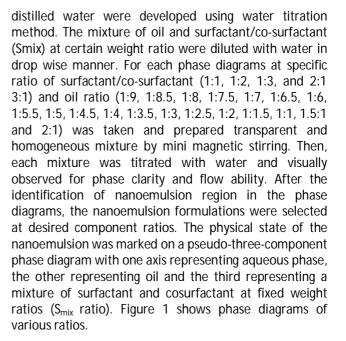
The solubility of Clopidogrel in various oils, surfactants, and cosurfactants was determined. Briefly, an excess amount of Clopidogrel (approximately 100 mg) was introduced into 1 gm of each vehicle, and mixture was kept in 20 ml beaker (covered with Aluminum foil).The mixture was stirred using mini magnetic stirrer (DBK Instruments) for up to the saturation point in beaker at 37°C. Samples were kept aside and allow standing for 48hr at ambient temperature to attain equilibrium. The equilibrated sample was centrifuged at 2,000 rpm for 10 min to remove the un-dissolved drug. Aliquot of 0.2ml was taken from clear supernatant. Filtered using membrane filter (0.45µm). The concentration of Clopidogrel was then quantified using U.V spectrophotometer (Systronics 2201). Table 1 shows the outcomes of the solubility studies.

Table 1: Solubility of Clopidogrel in Various LipidExcipients

Solubility of	CLP in oils	Solubility of CLP in Surfactants			
Oil (1 gm)	Solubility (mg/gm)	Surfactant (1 gm)	Solubility (mg/gm)		
Gelucire 44/14	2.9185	Labrafil M 2125	7.7927		
Capmul MCM NF	0.4170	Acrysol K 150	8.9970		
Sunflower oil	0.5788	Acrysol EL 135	14.6385		
Linseed oil	0.2964	Acconon MC8-2	42.2835		
Soyabean oil	1.1498	Tween 80	0.6170		
Olive oil	0.0658	Tween 20	14.902		
Solubility o co-surfactar		Solubility of CLP in co-solvents (1gm)			
Captex 200	0.0322	Capryol 90	0.4438		
PEG 400	32.951	PEG 400	32.9514		
Labrafac PG	0.0157	Propylene Glycol	0.0294		

Construction of Pseudo Ternary Phase Diagram Study⁷

On the basis of the trial batches of excipients, the pseudoternary phase diagrams of oil (Capmul MCM NF), surfactant: co-surfactant (Acrysol K 150: PEG 400), and



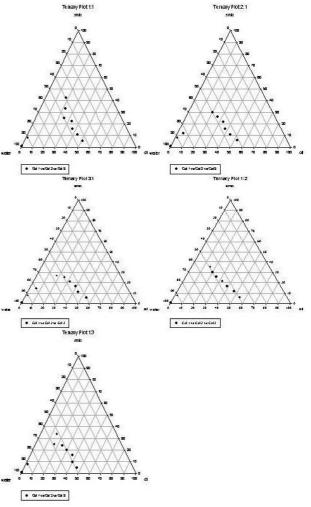


Figure 1: Ternary Phase Diagram Study on clopidogrel SNEDDS

Drug loading capacity

This study was done with the purpose to achieve the highest drug loading capacity in the formulation specifically for high dose drug molecules. Make 1 gm



SMEDDS formulation as per 1:1 smix and 8:2 Smix: oil (in 3 beaker). In all three beakers add 50 mg,100mg and 150 mg drug respectively and after complete dissolution of the drug add slowly water in it (up to 100 ml) and cover it with aluminum foil. Keep it for 24 hrs to check the reprecipitation of drug.

Table 2: Different batches for Optimization

Ingradiant	F1	F2	F3	F4	F5	F6	F7	F8
Ingredient	Qty. (mg)							
Drug : CLP	150	150	150	150	150	150	150	150
Acrysol K 150	400	450	540	600	600	680	220	300
PEG 400	400	450	260	300	200	220	660	600
Capmul MCM NF	200	100	200	100	200	100	100	100
Total	1150	1150	1150	1150	1150	1150	1150	1150

Preparation of liquid SNEDDS formulations

Sequences of SNEDDS were prepared in trial formulations with varying ratio of oil, surfactant, co-surfactant, and Clopidogrel. CLP concentration remains same in all formulations. All 8 formulations were prepared by taking accurate quantity of all excipients along with drug and stirred well till 10-15 min at 25°C in a small beaker. Heat the solution if drug don't get solubilize. Finally, the

mixture was kept at room temperature quantities of various formulations were depicted in table 3.

Table 3: Dilution	study of Clopidogrel SNEDDS
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Vehicles	I.	Ш	III	
Distilled water	Stable	Hazy within 2 Hrs	Hazy within 1 Hrs	
pH 1.2	Stable	Hazy within 6 Hrs	Hazy within 3 Hrs	
pH 6.8	Stable	Hazy within 3 Hrs	Hazy within 2 Hrs	

Formulation I: 75 mg Clopidogrel, 17.39 % Capmul MCM NF, 34.78 % Acrysol K 150, 34.78% PEG 400

Formulation II: 75 mg Clopidogrel, 17.39 % Capmul MCM NF, 46.96 % Acrysol K 150, 22.66% PEG 400

Formulation III: 75 mg Clopidogrel, 17.39 % Capmul MCM NF, 61.15 % Acrysol K 150, 17.39% PEG 400

Dilution Study

Dilution may better mimic conditions in the stomach following oral administration of SMEDDS pre-concentrate. Dilution study was done to access the effect of dilution on SMEDDS pre-concentrates. An entire 75 mg of Clopidogrel integrated in SNEDDS formulation.1 part SNEDDS of each solution was diluted with 10 parts of distilled water, Phosphate buffer pH 1.2 and Phosphate buffer pH 6.8 and observed. Observation of dilution studies is shown in Table 4.

	Effect of Temperature on Phase Separation, Flocculation, Precipitation							
Code After 4 week		After 8 week		After 12 week		Dispersibility	Inference	
oouc	2-8 °C	Room temperature	2-8° C	Room temperature	2-8 °C	Room temperature	study	interence
F1-F8	Not seen	Not seen	Not seen	Not seen	Not seen	Not seen	Grade A	Pass
	Centrifugation stability data (Phase Separation)							
	After	After 1 month After 2 month		After 3 month				
F1-F8	No	Not seen Not seen		Not seen		Grade A	Pass	
	Heating and cooling cycle(Creaming or Cracking)							
	After 12 hr After 24 hr		After 48 hr					
F1-F8	Not seen Not seen		Not seen		Grade A	Pass		
	Freeze thaw cycle (Phase Separation)							
	at	: -12 °C	at 5 °C		at 25°C			
F1-F8	No	ot seen	Not seen		Not seen		Grade A	Pass

SNEDDS characterization

Thermodynamic stability studies⁸

- ✓ Heating cooling cycle: Six cycles between refrigerator temperature (4°C) and 45°C with storage at each temperature of not less than 48 h was studied. Stable formulations at these temperatures were subjected to centrifugation test.
- ✓ Centrifugation: Accepted formulations were centrifuged at 2500 rpm for 40 min. Those formulations having absence of any phase separation were taken for the freeze thaw stress test.
- ✓ Freeze thaw cycle: Three freeze thaw cycles amongst -21 °C and +25 °C with storage at each temperature for not less than 48 h was done for the formulations.

Formulations which passed these thermodynamic stress tests were further taken for the dispersibility test for assessing the efficiency of self-emulsification.

Dispersibility test

The efficiency of self-emulsification of oral nanoemulsion was assessed using a standard USP dissolution apparatus type 2.⁸ 1 ml of each formulation was added to 500 ml of water at 37 \pm 0.5°C. A standard stainless steel dissolution paddle rotating at 50 rpm provided gentle agitation. The



in vitro performance of the formulations was visually assessed using the following grading system:

Grade A: Rapidly forming (within 1 min) nanoemulsion, having a clear or bluish appearance.

Grade B: Rapidly forming, slightly less clear emulsion, having a bluish white appearance.

Grade C: Fine milky emulsion that will form within 2 min.

Grade D: Dull, grayish white emulsion having slightly oily appearance that is slow to emulsify (longer than 2 min).

Grade E: Formulation, exhibiting either poor or minimal emulsification with large oil globules present on the surface.

Percentage Transmittance (λ max 650 nm)

The CLP SNEDDS were equipped by taking 100 times dilution of various media in 1 gm drug and the resulting nanoemulsion was observed visually for any turbidity. Subsequently, its % transmittance was measured at 650 nm using UV-vis spectrophotometer (systronics 2201). The studies were conducted by taking distilled water; phosphate buffer pH 1.2 and phosphate buffer pH 6.8.

Cloud point measurement study ¹²

The cloud point is a crucial feature in the SNEDDS consisting of non-ionic surfactants, and it is responsible for the successful formation of a stable microemulsion. The SNEDDS were compared for cloud point value. Each formulation was diluted with distilled water in the ratio of 1:250 and placed in a water bath with gradual increase in temperature. The point at which cloudiness occur was noted as cloud point.

Determination of droplet size/distribution and zeta-potential

Clopidogrel SNEDDS concentration (approximately 100 mg) was diluted with purified water (100mL) and gently shaken in a volumetric flask at 25°C. The droplet size/distribution and zeta-potential were analyzed by dynamic light scattering technique using a Zetasizer (Nano ZS, Malvern Instruments, UK) equipped with a 4.0 mW He-Ne red laser (633nm).

Drug diffusion studies

SNEDDS of CLP was filled in dialysis bag. *In vitro* release profile of SNEDDS was studied using USP apparatus type II at 37 ± 0.50 °C with a rotating speed of 100 rpm in dissolution media namely, pH 1.2 and 6.8 buffer so as to evaluate the effect of pH on *in vitro* dissolution. During the study, 3 ml of aliquots were removed at predetermined time intervals (5, 10, 15, 20, 25, 30, 40, 50 up to 90 min) from the dissolution medium and replaced with fresh buffer. The amount of CLP released in the dissolution medium (figure 3) was determined by UV spectrophotometer at λ max = 235 nm.

RESULTS

Pseudo-ternary phase diagrams study

The pseudo-ternary phase diagrams were mapped with the water titration method to identify the area of microemulsion regions at 37°C. The distilled water was used as diluting medium and added into the formulation. The proper ratio of one excipient to another in the SNEDDS formulation was analysed. Several formulations with different oil and Smix values (the ratio of surfactant to cosurfactant) were dispersed with water at 37°C. The pseudo-ternary phase diagrams of the formulation composed of Acrysol K 150, Capmul MCM NF and PEG 400 are shown in Figure 1. From figure, it was concluded that emulsifying zone is observed in almost all the ratios applied and no significant effect on Smix: oil ratio. The shadow area represents the o/w microemulsion existence region. The size of the nanoemulsion region in the diagrams was compared as, larger the size the greater the self- nanoemulsification efficiency.

Drug loading capacity

Capacity of drug engulfment in nanoemulsion was observed by putting three different quantities and concludes that there is no precipitation of drug even after seven weeks and shows his maximum drug loading capacity by appealing the industrial use of SNEDDS.

Preparation of liquid SNEDDS formulations

In following experiment we undergo various trial batches so as to explore the perfect combination of excipients. After various trials prepared on the basis of solubility of drug, final batches were prepared through ternary phase diagram study shown in table 2.

Dilution Study

A right blend of emulsifier is necessary for the development of SNEDDS formulation; to form stable Nanoemulsion. When 1 part SNEDDS of each solution was diluted with 10 parts of distilled water, HCl buffer 1.2 pH and phosphate buffer 6.8 pH (Table 3). It implies that the formulation was more stable because there was no precipitation.

SNEDDS Characterization

Thermodynamic stability studies

Various thermodynamic stability studies were performed by evaluating its temperature, centrifugation and dispersibility potential. A study reveals that there is no effect of any of the parameter on the transparency of nanoemulsion and not show any precipitation, phase separation, creaming or cracking given in Table 4.

Percentage Transmittance

Percentage transmittance of nanoemulsion in distilled water, HCl buffer pH 1.2 and Phosphate buffer pH 6.8 was 99%, 98% and 99 %, respectively, which was nearer to 100%. It indicates clear nanoemulsion was formed from the SNEDDS up to 100 times dilution with distilled water.



Cloud point measurement study

The cloud point is the temperature above which a clear formulation turns cloudy. At temperatures higher than the cloud point, an irreversible phase separation occurs due to dehydration of its ingredients, which may affect drug absorption. Hence, to avoid this phenomenon, the cloud point for SNEDDS should be above body temperature (37 °C). The cloud point for CLP SNEDDS was much higher (60-65 °C) which indicates that it will form stable nanoemulsion at physiological temperature i.e. in vivo, without risk of phase separation.

Determination of droplet size/distribution and zeta-potential

Particle size after nanoemulsification was the most important property of SMEDDS. Mechanisms of particle size effect on drug absorption may include improved release and facilitated lymphatic transport. The CLP SNEDDS showed somewhat parallel mean globule size within range of 16-42 nm when diluted with distilled water. The time required for formation of nanoemulsion after dilution with distilled water was just 38 second. The resultant nanoemulsion were transparent in appearance and they didn't show any symptom of phase separation and drug precipitation even after 24 h .Droplet size of optimal formulation was given in figure 2.

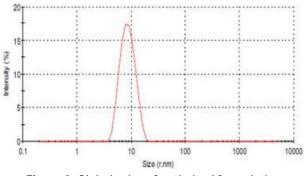


Figure 2: Globule size of optimized formulation

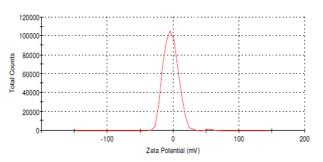


Figure 3: Zeta Potential of optimized formulation

Zeta potential of the SNEDDS is helpful to recognize the charge of oil globules in the emulsion. The increase in electrostatic repulsive forces between the globules averts the coalescence of nanoemulsion. In contrast, decrease in electrostatic repulsive forces can cause phase separation. Several studies have reported that the zeta potential played an important role in the interactions with mucus of the gastrointestinal tract. The zeta potential of all the batches lied between-0.366 to 4.51. The zeta potential of

the optimized formulation obtained by diluting CLP SNEDDS with distilled water (100 times) was -3.25 ± 0.62 mV and results are shown in Fig.3. The charge on an oil globule may be negative due to surfactants and/or co-surfactant present in the formulation.

Drug diffusion studies

Diffusion patterns of CLP from plain drug, marketed tablet formulation and various batches after reconstitution with HCl buffer (pH 1.2) and phosphate buffer (pH 6.8) are depicted in Fig.4 The patterns disclose that release of CLP from various batches varied with change in pH. It was observed from optimized batch that more than 60% drug released within 5 min and complete release was obtained within 15 min in pH 1.2 buffers, while 92% drug was released in 15 min in phosphate buffer (pH 6.8). This difference may be due to the difference in solubility of CLP at different pH. On the other hand, the release of CLP from the marketed tablet (clopilet) was only 72% within 90 min in pH 1.2 and pH 6.8. These results were found to be more sliding with plain drug having release of 56% within 90 min. Thus, we can say that the formulated SNEDDS having much potential than the plain drug and even from marketed dosage form and shows uniform in vitro release throughout the GIT irrespective of pH variations.

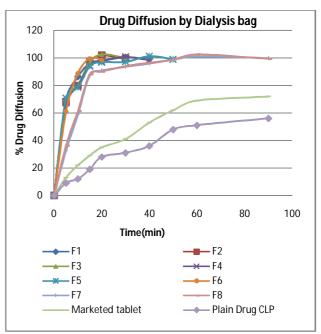


Figure 4: *In vitro* drug diffusion of formulation F1-F8 with plain drug and conventional tablet of CLP

CONCLUSION

The present research work could be summarized as successful development of SNEDDS of Antiplatelet CLP using Acrysol K150 (surfactant), Capmul MCM NF (oil), PEG 400 (Cosurfactant/Cosolvent) which gives stable and transparent nanoemulsion. Based on higher solubility, ultimate nanoemulsifying zone, lesser globule size with minimum polydispersity index, acceptable globule charge, higher transmittance, lower viscosity, higher cloud point



and superior drug release illustrate the potential use of Clopidogrel SNEDDS orally. Formulation can also be given in liquid solution or in capsule dosage form deemed to be the efficacious and patient compliant delivery system. Studies also showed how nanoemulsion formulation can be supportive for the delivery of hydrophobic compounds with capacity to load higher drug concentration, minimum surfactant concentration and proper infinite dilution without attaining drug precipitation.

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REFERENCES

- Shu-xia Cui, Shu-fang Nie, Li Li, Chang-guang Wang, and Wei-san Pan., "Preparation and Evaluation of Self-Microemulsifying Drug Delivery System Containing Vinpocetine" Drug Development and Industrial Pharmacy, 35, 2009, 603–611.
- R.P. Dixit, M.S. Nagarsenker., "Self-nanoemulsifying granules of ezetimibe: Design, optimization and evaluation" European journal of pharmaceutical sciences, 35, 2008, 183–192.
- Emad B. Basalious, Nevine Shawky, Shaimaa M. Badr-Eldin., "SNEDDS containing bioenhancers for improvement of dissolution and oral absorption of lacidipine. I: Development and optimization" International Journal of Pharmaceutics, 391, 2010, 203–211.
- Hong, J.Y., Kim, J.K., Song, Y.K., Park, J.S., Kim, C.K., "A new self emulsifying formulation of Itraconazole with improved dissolution and oral absorption. J.Control. Rel. 110, 2006, 332–338.
- Savi, P; Zachayus JL; Delesque-Touchard N; "The active metabolite of Clopidogrel disrupts P2Y12 receptor oligomers and partitions them out of lipid rafts". Proceedings of the National Academy of Sciences of the USA, 103, 29, 11069–11074
- 6. R.P. Dixit, M.S. Nagarsenker., "Self-nanoemulsifying granules of ezetimibe: Design, optimization and evaluation" European journal of pharmaceutical sciences, 35, 2008, 183–192.
- Shu-xia Cui, Shu-fang Nie, Li Li, Chang-guang Wang, and Wei-san Pan., "Preparation and Evaluation of Self-Microemulsifying Drug Delivery System Containing Vinpocetine" Drug Development and Industrial Pharmacy, 35, 2009, 603–611.
- Deepa Patel and Krutika K. Sawant, "Oral Bioavailability Enhancement of Acyclovir by Self-Microemulsifying Drug Delivery Systems (SMEDDS)" Drug Development and Industrial Pharmacy, 33, 2007, 1318–1326.

- 9. Anuradha Verma, Manish K.Singh, Babita Kumar, "Development and Characterization of Flutamide Containing Self-Microemulsifying Drug Delivery System (SMEDDS)", Int J Pharm Pharm Sci, Vol 3, Suppl 4, 60-65.
- 10. Sheikh Shafiq, Faiyaz Shakeel, Sushma Talegaonkar and Roop K. Khar, "Development and bioavailability assessment of ramipril nanoemulsion formulation" European Journal of Pharmaceutics and Biopharmaceutics, 66, 2007, 227–243.
- 11. Lanlan Wei, Peinan Sun, Shufang Nie and Weisan Pan, "Preparation and Evaluation of SEDDS and SMEDDS Containing Carvedilol", Drug Development and Industrial Pharmacy, 31, 2005, 785–794.
- Maulik J. Patel, Natvarlal M. Patel, Ritesh B. Patel, Rakesh. P. Patel, "Formulation and evaluation of selfmicroemulsifying drug delivery system of lovastatin" Asian Journal of Pharmaceutical Sciences, 2010, 5 (6), 266-275.
- Prabagar Balakrishnan, Beom-Jin Lee and Dong Hoon Oh, "Enhanced oral bioavailability of dexibuprofen by a novel solid Self-emulsifying drug delivery system (SEDDS)" European Journal of Pharmaceutics and Biopharmaceutics, 72, 2009, 539–545.
- 14. Ping Zhang, Ying Liu, Nianping Feng and Jie Xu, "Preparation and evaluation of self-microemulsifying drug delivery system of oridonin" International Journal of Pharmaceutics 355, 2008, 269–276.
- Shweta Gupta, Sandip Chavhan, Krutika K. Sawant., "Selfnanoemulsifying drug delivery system for adefovir dipivoxil: Design, characterization, *in vitro* and ex *vivo* evaluation" Colloids and Surfaces A: Physicochem. Eng. Aspects, 392, 2011, 145–155.
- Wei Wu, Yang Wang, Li Que, "Enhanced bioavailability of silymarin by self-microemulsifying drug delivery system" European Journal of Pharmaceutics and Biopharmaceutics, 63, 2006, 288–294.
- 17. Sagar D. Mandawgade, Shobhona Sharma and Sulabha Pathak, "Development of SMEDDS using natural lipophile: Application to _-Artemether delivery" International Journal of Pharmaceutics, 362, 2008, 179–183.
- Xuemei Wu, Jianhua Xu, Xiuwang Huang and Caixia Wen, "Self-microemulsifying drug delivery system improves curcumin dissolution and bioavailability" Drug Development and Industrial Pharmacy, 2011, 37(1), 15–23.
- 19. HaiRong Shen and MingKang Zhong, "Preparation and evaluation of self-microemulsifying drug delivery systems (SMEDDS) containing atorvastatin" JPP 2006, 58, 1183– 1191.
- Abhijit A. Date, M.S. Nagarsenker, "Design and evaluation of self-nanoemulsifying drug delivery systems (SNEDDS) for cefpodoxime proxetil" International Journal of Pharmaceutics, 329, 2007, 166–172.

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