



Phytochemical Analysis of *Didymocarpus pedicellatus*

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ABSTRACT

Phytochemical analysis of extract of *Didymocarpus pedicellatus* was conducted to detect the presence of various secondary metabolites. These plants are widely used in South Indian Cuisine. It is commonly known as Black stone flower. *Didymocarpus pedicellatus* has a strong ethanopharmaceutical background. *D. pedicellatus* revealed the presence of Alkaloids, flavonoids, tannis, phenols etc.

Keywords: *Didymocarpus pedicellatus*, Phytochemical extraction, Secondary metabolites.

INTRODUCTION

Indian spices have many medicinal properties. *Didymocarpus pedicellatus* commonly called as Kalpasi in Tamil is widely used in Indian Cuisine. *Didymocarpus pedicellatus* (Kalpasi) is a valuable although a lesser known medicinal plant. It is popularly known as black stone flower and is a rare flower in the family Gesneriaceae. The plant is native to Tropical Asia.¹ In Ayurveda it is known as shilapushpa, shantapushpi and sometimes pasanbheda.² It is often dried and used as an Indian spice. Black stone flowers have many medicinal properties. They are a good pain reliever and also promote early healing of wounds. It helps in digestion and suppresses the respiratory disorders. In this study we are trying to find the phytochemical properties from the dried flowers of *Didymocarpus pedicellatus*.

Traditionally *D. pedicellatus* used in the treatment of renal diseases particularly kidney stones.³ It also regulates calcium absorption in the body. The plant is known for its diuretic effect and in maintaining healthy urinary tract.⁴

Phytochemical studies have attracted the attention of plant scientists due to the development of new sophisticated techniques. These techniques played a significant role in giving the solution to systematic problems on the one hand and in the search for additional resources of raw materials for pharmaceutical industry on the other hand.⁵

Medicinal plants contain some organic compounds which provide definite physiological action on the human body and these bioactive substances include tannins, alkaloids, carbohydrates, Terpenoids, steroids and flavonoids.^{6,7}

D. pedicellatus extract was found to possess a high content of total polyphenolics, exhibit potent reducing power and significantly scavenge free radicals including several reactive oxygen species (ROS) and reactive nitrogen species (RNS). The extract also significantly and dose-dependently protected against Fe-NTA plus H₂O₂-mediated damage to lipids and DNA.⁸

MATERIALS AND METHODS

Preparation of plant extract

Dried flowers of *Didymocarpus pedicellatus* were collected and homogenized into a fine powder using mechanical grinder.

Preparation of Extracts

Powdered plant material was subjected to successive solvent extract using solvents like Ethanol, Methanol, Chloroform, and Acetone. 20g of powdered flower sample was added to 200 ml of suitable solvents and the crude was extracted after 48 hrs. The extracts were filtered and evaporated. Phytochemical analyses were carried out from the crude.

Phytochemical analysis

Phytochemical analysis were done by certain reagents to test the alkaloid, carbohydrates, reducing sugar, flavonoids, glycosides, phenolic compounds, saponins, steroids, amino acids and tannins.

Test for Alkaloids

Extracts were mixed with dilute hydrochloric acid and following tests were done.

a. Mayer's Test

Extracts were treated with Mayer's Reagent (Mercuric chloride & Potassium iodide). Formation of a cream indicates the presence of alkaloids.

b. Wagner's Test

Extracts were treated with Wagner's Reagent (Iodine & Potassium iodide). Formation of a reddish brown precipitate indicates the presence of alkaloids.

Test for Flavonoids

Aqueous extracts was treated with dilute ammonia and conc. Sulphuric acid was added. Yellow colouration indicates the presence of flavonoids.

Test for Terpenoids

To the extracts 2ml of chloroform was added, conc. Sulphuric acid was carefully added to form a layer. Reddish brown colouration at the interface indicates the presence of terpenoids.

Test for Steroids

a. Extracts were mixed with 2ml of chloroform and conc. sulphuric acid was added sideways. A red colour producing at the lower chloroform layer indicate the presence of steroids.

b. Extracts were mixed with 2ml of chloroform, 2 ml of conc. sulphuric acid and 2ml of acetic acid were poured into the mixture. Development of greenish colouration indicates the presence of steroid.

Test for Phenols

Ferric chloride Test

Extracts were treated with 3 to 4 drops of ferric chloride solution. Formation of bluish black colour indicates the presence of phenol.

Test for Glycosides

a. Libermann Burchard's Test

Extracts were treated with 2 ml of chloroform and 2 ml of acetic acid. The mixture was cooled in ice and conc. Sulphuric acid was added. Colour change from violet to blue to green the indicates the presence of glycosides

b. Salkowski's Test

Extracts were mixed with 2ml of chloroform. Then 2 ml of conc. Sulphuric acid was added carefully and shaken gently. A reddish brown colour indicates the presence of steroidal ring.

c. Keller-kilani Test

Extracts were mixed with 2ml of glacial acetic acid containing 1 to 2 drops of ferric chloride solution. The mixture was then poured into another test tube containing 2 ml of conc. Sulphuric acid. A brown ring at the interface indicates the presence of cardiac glycosides.

Test for Protein

a. Ninhydrin

Extracts were boiled with 2 ml of ninhydrin solution. Appearance of violet colour indicates the presence of proteins.

b. Xanthoproteic

Extracts were treated with few drops of conc. Nitric acid. Formation of yellow colour indicates the presence of proteins.

Test for Carbohydrates

a. Benedict's solution

Extracts were treated with 2 ml of Benedict's reagent and gently heated. Formation of orange red precipitate indicates the presence of reducing sugar.⁹

b. Fehling's Test

Equal volume of Fehling A and Fehling B reagents were mixed together and 2 ml of it was added to the extract and gently boiled. A brick red colour indicates the presence of reducing sugar.

c. Iodine Test

Extracts were mixed with 2 ml of Iodine solution, a dark blue or purple colour indicates the presence of Carbohydrates.

Test for Tannins

Extracts were boiled with 10 ml of water in a test tubes. A few drops of ferric chloride was added and observed for blue black colouration.

Test for Saponins

Extracts were boiled with distilled water. The solution was shaken vigorously for a stable, persistent, froth or foam.

Table 1: Phytochemical Screening of *Didymocarpus pedicellatus*

Tests	Ethanollic Extract	Acetone Extract	Methanol Extract	Chloroform Extract
Alkaloids				
Mayer's test	+	+	+	+
Wagner's Test	+	+	+	+
Glycosides				
Liberman's Test	+	+	-	-
Salkowski's Test	+	+	+	+
Keller Kilani's Test	+	+	+	+
Carbohydrates				
Fehling's Test	+	+	+	+
Benedict's Test	+	+	+	+
Iodine Test	+	+	+	+
Proteins				
Xanthoproteic Test	+	+	+	+
Ninhydrin Test	+	+	+	+
Tannins	+	-	+	+
Flavonoids	+	+	+	+
Steroids	+	-	+	+
Terpenoids	+	+	+	+
Phenol	+	+	+	+
Saponin	+	+	+	+

RESULTS AND DISCUSSION

The phytochemical active compounds of *Didymocarpus pedicellatus* were qualitatively analyzed for flowers and the results are presented.

The phytochemical studies of four solvents shows the presence of flavonoids, steroids, terpenoids, saponins, tannins, proteins, phenols, alkaloids, glycosides and carbohydrates in almost all the four extracts. Different phytochemical have been found to possess a wide range of activities, which may help in protection against chronic diseases. *D. pedicellatus* contains Chalcones¹⁰, Polyterpenes like didymocarpol, didymacarpinol.⁴ Since *D. pedicellatus* contains flavonoids, it may be also valued for anticancer activity.

Results obtained from phytochemical screening were shown in Table.1. Thus the present investigation revealed that *D. pedicellatus* appears to be rich in secondary metabolites, the preliminary study shows the presence of bioactive compounds. Further studies are in progress to isolate the active compounds.

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